We are now in a position to determine the volume of fluid $\nabla_{0}$ that must exit the compression chamber via the knock-off tube in order for the pressure within the compression chamber to decay to ambient conditions in accordance with the postulated pressure-time function given by (4). We write

$$
\begin{equation*}
V_{0}=\int_{0}^{\infty} Q d t \tag{51}
\end{equation*}
$$

Since the volume rate of flow $Q$ diminished in an exponential fashion, the limits of integration on the time $t$ are zero and infinity.

From the combination of equations (50) and (51), we have

$$
\begin{equation*}
V_{0}=\frac{4 \pi P g}{\rho L} \sum_{j=1}^{\infty}\left[\frac{1}{\Theta_{1}}-v \lambda_{j}^{2}\right]\left(\frac{1}{\lambda_{j}^{3}}\right) \int_{0}^{\infty}\left[e^{-\lambda_{j}^{2} v t}-e^{-t / \oplus_{1}}\right] d t \tag{52}
\end{equation*}
$$

from which

$$
\begin{equation*}
V_{0}=\frac{4 \pi P_{g}{ }_{1}}{\mu L} \sum_{j=1}^{\infty} \frac{I}{\lambda_{j}^{4}} \tag{53}
\end{equation*}
$$

Recalling that $\lambda_{j}$ is defined by equation (45), we can obtain from reference (e)

$$
\begin{aligned}
& \lambda_{1} R_{0}=2.4048 \\
& \lambda_{2} R_{0}=5.5201 \\
& \lambda_{3} R_{0}=8.5537 \\
& \lambda_{4} R_{0}=11.7915 \\
& \text { etc. }
\end{aligned}
$$

Thus equation (53) becomes

$$
\begin{equation*}
V_{0} \approx \frac{\pi \Theta_{1} P_{g} R_{0}^{4}}{8 \mu L} \tag{55}
\end{equation*}
$$

The time constant $\Theta_{1}$ can be found from equation (55) providing $V_{0}$ is known. It is understood that the pressure in the compression chamber is increased by forcing an additional quantity of fluid into the chamber. From reference ( $f$ ) the compressibility" of the subject fluids (SAE 10 and SAE 20 oil) can be closely approximated from the following equation

$$
\begin{align*}
\frac{V_{1}}{V_{2}}= & 1.00-\left(4.31 \times 10^{-8}\right) \mathrm{P}_{\mathrm{g}}+\left(5.51 \times 10^{-11}\right) \mathrm{P}_{\mathrm{g}}^{2}  \tag{56}\\
& -\left(5.03 \times 10^{-18}\right) \mathrm{P}_{\mathrm{g}}^{3}
\end{align*}
$$

where the nomenclature is
$V_{1}$. . . volume of fluid under pressure $P_{g}$ (volume of compression chamber), in ${ }^{3}$
$\nabla_{2}$. . . volume of fluid under atmospheric pressure (volume that would create pressure $\mathrm{P}_{\mathrm{g}}$ if compressed to volume $\mathrm{V}_{1}$ ), in ${ }^{3}$

Since $V_{0}=V_{2}-V_{1}$, then
$V_{0}=V_{1} P_{g}\left[\frac{\left(4.31 \times 10^{-8}\right)-\left(6.51 \times 10^{-12}\right) \mathrm{P}_{\mathrm{g}}+\left(5.03 \times 10^{-18}\right) \mathrm{P}_{\mathrm{g}}^{2}}{1.00-\left(4.31 \times 10^{-6}\right) \mathrm{P}_{\mathrm{g}}+\left(6.51 \times 10^{-12}\right) \mathrm{P}_{\mathrm{g}}^{2}-\left(5.03 \times 10^{-16}\right) \mathrm{P}_{\mathrm{g}}^{3}}\right](57)$

To simplify the writing of (57), we introduce the term $K_{p}$, defined as follows for the subject fluids

